

Faculty of Science Master of Science (M.Sc.) (W. E. F.: 2023-24) Document ID: SUTEFSCM-01

Name of Faculty	:	Faculty of Science
Name of Program	:	Master of Science
Course Code	:	2MSO05
Course Title	:	Practical – I (Organic chemistry and Inorganic Chemistry)
Type of Course	:	Basic Science
Year of Introduction	:	2023-24

Prerequisite	:	Basic practical knowledge of Inorganic and organic chemistry.
Course Objective	:	Chemistry Practicals provide students to learn about synthesis, separation, purification and identification of organic as well as Inorganic compound. It develops the ability to correlate the chemical and physical properties of elements and their compounds.
Course Outcomes	:	At the end of this course, students will be able to:
	CO1	Analyze separation, purification and identification of ternary mixtures by Chemical and physical methods
	CO2	Get awareness about laboratory safety and handling of chemicals
	CO3	Apply knowledge to develop method for qualitative identification elements from the mixture having applications in industry and research.
	CO4	Understand importance of metal complexes and green methods for the synthesis.

#### **Teaching and Examination Scheme**

Teaching Scheme (Contact		Credits	Examination Marks					
	Hours)			Theory	Marks	Practica	l Marks	Total
L	Т	Р	С	SEE	CIA	SEE	CIA	Marks
0	0	8	4	-	-	70	30	100

Legends: L-Lecture; T-Tutorial/Teacher Guided Theory Practice; P – Practical, C – Credit, SEE – Semester End Examination, CIA - Continuous Internal Assessment (It consists of Assignments/Seminars/Presentations/MCQ Tests, etc.)

NOTE: This specification table shall be treated as a general guideline for the students and the teachers. The actual distribution of marks in the question paper may vary slightly from above table



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# Suggested List of Experiments/Tutorials

Sr. No.	Name of Experiment/Tutorial	Teaching Hours	Mapping With COs
	Part A: Organic Chemistry Practical		
	Qualitative analysis and Separation of given unknown		
	organic mixture (S+S+S) as well as (L+L+L) (in case of		
	liquid sepapration by physical and chemical method		
	both permitted)		
1	Acid+Base+Phenol	04	CO1
2	(Acid + Acid + Neutral)	04	CO2
3	(Acid + Base + Acid)	04	CO3
4	(Acid + Base + Base)	04	
5	(Base + Base + Neutral)	04	
6	(Acid + Neutral + Neutral)	04	
7	(Neutral + Base + Neutral)	04	
8	(Acid + Phenol + Acid)	04	
9	(Phenol + Base + Phenol)	04	
10	(Phenol + Phenol + Neutral)	04	
11	(Phenol + Neutral + Neutral)	04	
12	(Acid + Base + Phenol)	04	
	Part B: Inorganic Chemistry Practical		
	Preparation and determination of purity of double and		
	complex salts. At least ten preparations should be done.		
1	To Prepare Tetraamine copper sulphate from copper sulphate.	04	
2	To Prepare Tri(thiourea) cuprous sulphate from cuprous sulphate.	04	
3	To Prepare Tri(thiourea) cuprous chloride from cuprous sulphate.	04	CO3 CO4
4	To Prepare Hexaaminenickel(II) chloride from Nickel Chloride.	04	
5	To Prepare Potassium trioxalatochromate(III) from Potassium dichromate.	04	
6	To Prepare Potassium trioxalato Aluminate from Aluminum Sulphate.	04	
7	To Prepare Hexa(thiourea) plumbus nitrate from Lead	04	
8	To Prepare Hexaamino cobaltic chloride from CoCl <sub>2</sub> and NH <sub>4</sub> Cl.	04	
9	To Prepare Prussain Blue from Iron and Potassium ferrocyanide.	04	
10	To Prepare Penta thiourea dicuprous nitrate.	04	



Suggested Distribution of Theory Marks Using Bloom's Taxonomy						
Level	Remembrance	Understanding	Application	Analyse	Evaluate	Create
Weightage	25	25	25	25	-	-

# Major Equipment/ Instruments and Software Required

Sr. No.	Name of Major Equipment/ Instruments and Software
1	Fusion tube
2	Theil's tube
3	Capillary
4	Weighing Balance
5	Measuring cylinder
6	Conical flask
7	Glass rod
8	Beaker
9	Test tubes
10	Funnel
11	Test tube stand
12	China dish

### **Suggested Learning Websites**

Sr. No.	Name of Website
1	: https://vlab.amrita.edu/index.php?sub=2&brch=191∼=345&cnt=1
3	: <u>https://vlab.amrita.edu/index.php?sub=2&amp;brch=191∼=344&amp;cnt=1</u>

#### **Reference Books**

Sr. No.	Name of Reference Books
1	A text book of practical organic chemistry – A. I. Vogel
2	An Advance Course in practical Chemistry, A K. Nad, B. Mahapatra and A. Ghoshal.
3	A handbook of quantitative and qualitative analysis – H. T. Clarke
4	Comprehensive Practical Organic Chemistry: Preparations and Quantitative Analysis V K Ahluwalia
5	Practical organic Chemistry - Mann and Saunders.
6	An Advanced Course in Practical Chemistry
7	Advanced Practical Inorganic Chemistry
8	Vogel's Textbook of Quantitative Chemical Analysis
9	An Advanced Course in Practical Chemistry